



**CALIFORNIA STATE SCIENCE FAIR  
2002 PROJECT SUMMARY**

<b>Name(s)</b> <b>Michael R. Davis</b>	<b>Project Number</b> <b>S0406</b>
<b>Project Title</b> <b>The Effect of Temperature on the Enzyme Peroxidase</b>	
<p style="text-align: center;"><b>Abstract</b></p> <p><b>Objectives/Goals</b> I tested the effects of temperature on the enzyme peroxidase.</p> <p><b>Methods/Materials</b> In the presence of hydrogen peroxide, peroxidase reacts to produce visible gas bubbles. I used prepared slices of potato, with uniform size and shape, because they are a peroxidase source. The potatoes were heated/cooled to temperatures that ranged from one to one-hundred degrees Celsius. I used a pipette to apply 1 mL of hydrogen peroxide to the surface of the potato. Then I measured the area of peroxidase that reacted with the potato.</p> <p><b>Results</b> I found that the amount of foam increased relative to the temperature, until it reached 62° C, at which point it dropped off to 0%.</p> <p><b>Conclusions/Discussion</b> As the temperature increased, so did the amount of foam. The kinetic energy of the enzyme increased, enhancing the interaction with the substrate, causing a larger reaction. However, when the enzyme reached 62° C., it became denatured and useless.</p>	
<b>Summary Statement</b> I tested the effects of temperature on the enzyme peroxidase.	
<b>Help Received</b> I received no help on my project.	