

# CALIFORNIA STATE SCIENCE FAIR 2006 PROJECT SUMMARY

Name(s)

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**Project Number** 

S0517

## **Project Title**

# The Effect of Temperature on the Affinity of Carbon Dioxide to Molecular Sieve 4A

## **Abstract**

## Objectives/Goals

The purpose of my experiment is to find the optimum temperature for absorbing CO2 in molecular sieve 4A.

#### Methods/Materials

I first heat up my column to the test temperature. Then, I take a sample of 10% CO2 and 90% air with my input gas syringe. I then push this air through my column until there is no space left in the syringe. I simultaneously fill up the output syringe, #syringe 2,# to the same level that syringe 1 was at after filling it with air. I then heat up my oven to the maximum temperature and take a gas chromatograph sample of syringe 2. I then open the valves leading to the peristaltic pump and pump the gas in the column into syringe 3. I then take a sample of this gas and run it three times through my gas chromatograph.

#### Results

The data for sample 1 showed that the average percent in relation to the amount of air desorbed is 69.7% CO2 at 50°C. The data for sample 2 showed that the average percent of CO2 in the desorbed air is 74.6% CO2 at 100°C. The data for sample 3 showed that the average percent of CO2 in the desorbed air is 62.1% CO2 at 150°C. The data for sample 4 showed that the average percent of CO2 in the desorbed air is 64.4% CO2 at 200°C.

## **Conclusions/Discussion**

I can conclude that the optimum temperature for carbon dioxide absorption into molecular sieve 4A with minimal absorption of air is around 100°C. My Student#s t-Tests show that there is a significant difference between the data of each sample.

## **Summary Statement**

My project is about helping the problem of global warming by filtering CO2 from the atmosphere.

### Help Received

My dad supervised my setting up of a high pressure helium system and gas chromatograph use.