



**CALIFORNIA STATE SCIENCE FAIR
2017 PROJECT SUMMARY**

Name(s) Samantha C. Gaiera	Project Number J0904
Project Title Does Salinity Affect Ocean Acidification?	
<p style="text-align: center;">Abstract</p> <p>Objectives/Goals The purpose of this experiment was to investigate if freshwater or saltwater around the Earth was becoming more acidic due to increased carbon emission produced by humans.</p> <p>Methods/Materials A model was made to simulate earth's atmosphere and oceans. Distilled water with different salinity levels was exposed to CO₂ gas. The pH of the water was measured every five minutes with a digital pH meter for 30 minutes.</p> <p>Results The hypothesis that if the salinity of water increases and the solutions are exposed to CO₂ then the difference between initial and final pH of the solution will also increase was supported. The average change in pH was the highest with 36 ppt (1.33) and lowest with plain distilled water (1.0).</p> <p>Conclusions/Discussion This experiment is important because it shows that saltwater absorbs more CO₂ than freshwater. My research suggests that in places of higher salinity, CO₂ will be absorbed in greater amounts resulting in greater impact to those areas. As humans continue to increase carbon emissions, fresh water will be taking in less CO₂ than the ocean.</p>	
Summary Statement This experiment shows that as salinity of water increases, the amount of CO ₂ absorbed also increases.	
Help Received I designed and built the simulator by myself and my mother helped me collect materials.	