



**CALIFORNIA STATE SCIENCE FAIR  
2004 PROJECT SUMMARY**

<b>Name(s)</b> <b>Lauren M. Palumbi</b>	<b>Project Number</b> <b>S1417</b>
<b>Project Title</b> <b>The Effects of Runoff Toxic Levels on Strongylocentrotus purpuratus Embryos</b>	
<b>Objectives/Goals</b> The objective of this project is to assess the potential biohazard of runoff toxins. I measured the effects of Culpic Chloride, Zinc Chloride, and Lead Nitrate on the early development of sea urchins by utilizing bioassay techniques.	
<b>Abstract</b>	
<b>Methods/Materials</b> Methods 13 test tubes, Compound microscope, Fertilized embryos of the sea urchin species Strongylocentrotus Purpuratus, Solutions of Culpic Chloride at $3.15 \times 10^{-7}$ molar, $3.15 \times 10^{-6}$ molar, and $3.15 \times 10^{-5}$ molar, Solutions of Zinc Chloride at $3.82 \times 10^{-6}$ molar, $3.82 \times 10^{-5}$ molar, and $3.82 \times 10^{-4}$ molar, Solutions of Lead Nitrate at $1.69 \times 10^{-7}$ molar, $1.69 \times 10^{-6}$ molar, and $1.69 \times 10^{-5}$ molar, KCl solution, Syringe, Observation dishes for microscope, Plastic pipettes, A clock or stopwatch Procedure In holding with bio-assay procedure, sea urchin eggs were placed in a test tube of each dilution of the chemicals, including 3 controls. The eggs were then examined under a microscope at 1 hour, 4 1/2 hours, 5 1/2 hours, 17 hours, and 24 hours. The number of embryos dividing and their various stages of development was recorded.	
<b>Results</b> Of the three chemicals used to test toxicity, Lead and Copper were the ones to have the most drastic effect. Copper seemed to create the most disruption in the health of the embryos. This proves that increasing levels of Copper and Lead in runoff water could potentially prove to affect sea urchin populations and therefore disrupt other species dependent on them.	
<b>Conclusions/Discussion</b> The values used in this experiment were taken from the First Flush Report of 2000. Since that time the amount of Copper in the water has risen from $3.15 \times 10^{-7}$ molar to $4.7 \times 10^{-6}$ molar. In about another five years the amount of copper could rise to $1.5 \times 10^{-5}$ . That is roughly 953 ppb (parts per billion). It would take about 12 years at an annual increase of $2.2 \times 10^{-6}$ molar of copper to reach the most concentrated experimental copper value of $3.15 \times 10^{-5}$ or 2002 ppb. All of these values greatly surpass the background concentration of copper in sea water which is $3.1 \times 10^{-8}$ molar or 2 ppb.	
<b>Summary Statement</b> The purpose of this project is to asses the potential hazard of runoff water to local marine species and thier reproductive functions.	
<b>Help Received</b> Research done and lab equipment used at Hopkins Marine Station, Monterey. Equipment used under supervision of Dr. Stephen Palumbi. Assistance with board provided by Dr. Julie Alipaz, Katherine Gibson, and Chris Patton.	