



**CALIFORNIA STATE SCIENCE FAIR  
2013 PROJECT SUMMARY**

<b>Name(s)</b> <b>Ryan J. Bunk</b>	<b>Project Number</b> <b>S0694</b>
<b>Project Title</b> <b>Nanoporous Niobium Oxide</b>	
<b>Abstract</b> <b>Objectives/Goals</b> This project was to determine if photoelectrochemical etching ("photoetching") would successfully increase surface area of thin films by increasing nanoporosity. <b>Methods/Materials</b> Samples of niobium metal foil thicknesses 0.5-1mm thickness were heated until the oxide layer was lustrous dark blue, indicating thin film formation. 5 Samples were set aside as control, 8 samples were put in experimental. Experimental samples were photoetched under 1 molar sulfuric acid, ultraviolet C light, and a 1.5V electrical potential for 45 minutes. Control samples were not processed further after heating. Samples were analyzed by immersing in a solution of aqueous 70% isopropanol and elemental iodine, and determining quantity of solution absorbed by redox titration. Surface buildup of iodine solution by wetting was removed by pressing against the side of the beaker prior to titration. <b>Results</b> The samples that had been photoetched, on average, increased in solution absorption 172%, indicating that the samples had experienced an increase in nanoporosity. The samples were not discolored by etching, indicating that increases in macroporosity or other large material removal were negligible. <b>Conclusions/Discussion</b> The experimental, photoelectrochemically etched samples had successfully increased in nanoporosity, and by extension, increased in surface area.	
<b>Summary Statement</b> Increasing surface area of photocatalytic niobium V oxide thin films.	
<b>Help Received</b> Used lab equipment and materials at Adolfo Camarillo High School under the supervision of Mr. Tanner, Parents and Mr. Inouye reviewed and made suggestions on report, abstract, and poster	