



# CALIFORNIA SCIENCE & ENGINEERING FAIR 2019 PROJECT SUMMARY

<b>Name(s)</b> <b>William Porayouw</b>	<b>Project Number</b> <b>S0619</b>
<b>Project Title</b> <b>Constructing Earth-Abundant Core Shell Plasmonic Photocatalysts for Hydrogen Production via Water Splitting</b>	
<p style="text-align: center;"><b>Abstract</b></p> <p><b>Objectives</b> Currently, new alternatives for current greenhouse gases are of great interest, and hydrogen is a major player in the race for a new main source of fuel. Hydrogen can be produced through a photocatalytic water splitting process, and although titanium dioxide (TiO<sub>2</sub>) is a well known photocatalyst, factors such as its high recombination rate and an absence of a visible light absorption peak hinders its performance. By constructing transition metal-metal oxide core shell nanostructures (CSNs) that introduce copper (Cu) as a co-catalyst core, these issues with the semiconductor material can be addressed.</p> <p><b>Methods</b> To create efficient photocatalysts for hydrogen production, synthesis methods for earth abundant core shell Cu@TiO<sub>2</sub> nanostructures were designed. In order to construct these CSNs, a Cu core was synthesized through a modified sol-gel method with Hexamethylenediamine (HDA) as a capping agent and glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>) as a reducing agent, and size and morphology was controlled to create uniform nanostructures. Then, titanium isopropoxide (TIP) was hydrolyzed into TiO<sub>2</sub> with a diethanolamine (DEA) base catalyst, and coated on the Cu core with the assistance of HDA as a surfactant.</p> <p><b>Results</b> The CSNs were characterized through TEM and SEM imaging, UV-vis spectroscopy, and a photocatalytic test based on light irradiation. After 5 hours in a photocatalytic chamber, it was found that industrial grade TiO<sub>2</sub> produced the least amount of hydrogen, while prepared hollow TiO<sub>2</sub> produced twice as much as the industrial type, and Cu@TiO<sub>2</sub> produced the greatest amount of hydrogen, about 100 times as much as industrial TiO<sub>2</sub>.</p> <p><b>Conclusions</b> Cu successfully accelerated the semiconductor-light reactions by expanding the absorption spectra of TiO<sub>2</sub> and optimizing its plasmonic resonance property, lowering the band gap, and reducing recombination. In addition, Cu, is a cost-effective and earth-abundant potential co-catalyst, which makes the metal convenient for large scale manufacturing. Therefore, the Cu@TiO<sub>2</sub> CSNs proved to be efficient photocatalysts that should be further explored.</p>	
<b>Summary Statement</b> I constructed novel core shell nanostructures that utilized titania as a semiconductor shell and copper as a plasmonic core in order to optimize photocatalytic efficiency.	
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